

Chen-Yi Wang,<sup>a\*</sup> Ling-Xiang Guo,<sup>b</sup> Jian-Zhong Chen<sup>a</sup> and Ping Xia<sup>a</sup><sup>a</sup>Department of Chemistry, Huzhou University, Huzhou 313000, People's Republic of China, and <sup>b</sup>Department of Chemistry and Chemical Engineering, Southeast University, Nanjing 210096, People's Republic of China

Correspondence e-mail: chen\_yi\_wang2006@yahoo.com.cn

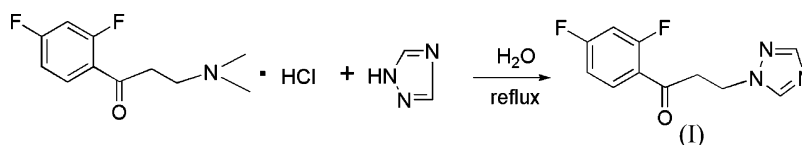
## Key indicators

Single-crystal X-ray study  
 $T = 294$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.041  
 $wR$  factor = 0.126  
Data-to-parameter ratio = 12.3For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## 1-(2,4-Difluorophenyl)-3-(1H-1,2,4-triazol-1-yl)propan-1-one

In the crystal structure of the title compound,  $\text{C}_{11}\text{H}_9\text{F}_2\text{N}_3\text{O}$ , dimers are formed *via*  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonding. These dimers are located around centres of inversion.Received 16 July 2006  
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## Comment

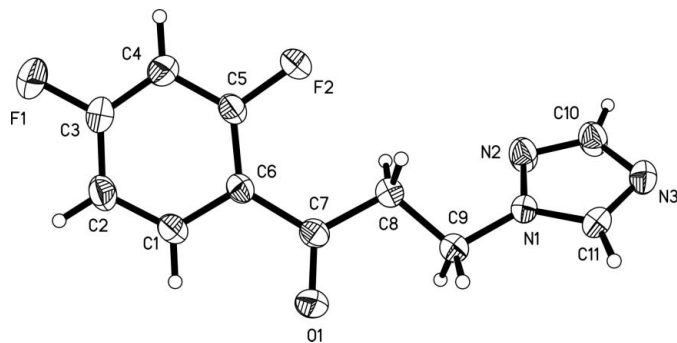
It is well known that compounds containing the triazole ring have good fungicidal and plant growth-regulating activities (Konosu *et al.*, 2001; Miyauchi *et al.*, 1995; Dyck *et al.*, 2003; Sekimata *et al.*, 2002; Ruel *et al.*, 2003). We report here the crystal structure of the title compound, (I) (Fig. 1).Bond lengths and angles of (I) are within normal ranges (Allen *et al.*, 1987) and are comparable with those in a related compound (Wan *et al.*, 2005). The molecule of (I) is non-planar, with the benzene and triazole rings making a dihedral angle of  $50.6(2)^\circ$ .The crystal structure of (I) is stabilized by an intermolecular  $\text{C11}-\text{H11}\cdots\text{O1}$  hydrogen bond [ $\text{H11}\cdots\text{O1} = 2.36$  Å,  $\text{C11}\cdots\text{O1} = 3.245(3)$  Å and  $\text{C11}-\text{H11}\cdots\text{O1}^i = 160^\circ$ ; symmetry code: (i)  $-x + 2, -y + 1, -z + 2$ ] (Fig. 2).

## Experimental

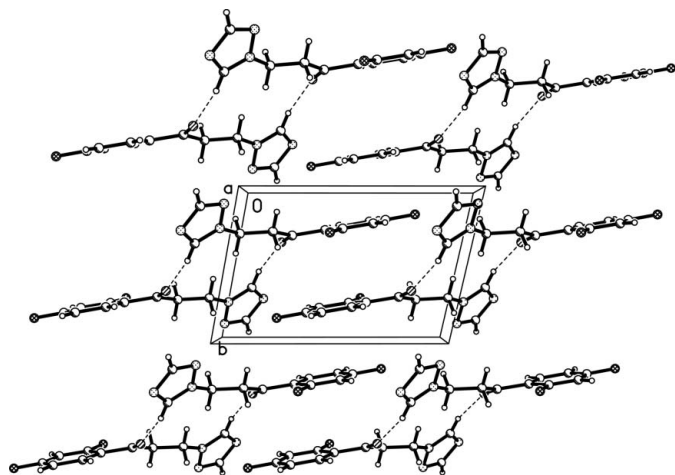
The title compound was prepared according to the method reported by Ehrhardt *et al.* (1982). 1H-1,2,4-Triazole (0.15 mol, 10.35 g) was dissolved in water (20 ml). The mixture was heated to reflux and kept at this temperature for 1 h, and then a solution of 1-(2,4-difluorophenyl)-3-(dimethylamino)propan-1-one hydrochloride (0.14 mol, 34.9 g) in water (45 ml) was added dropwise over a period of 2 h to the above heated mixture. After the addition, the reaction mixture was stirred at reflux for a further 2 h and then cooled to room temperature. A white solid was collected, which was recrystallized from ethyl acetate to give colourless crystals of (I) (yield 82%).

## Crystal data

 $\text{C}_{11}\text{H}_9\text{F}_2\text{N}_3\text{O}$   
 $M_r = 237.21$   
Triclinic,  $P\bar{1}$   
 $a = 6.939(3)$  Å  
 $b = 7.743(4)$  Å  
 $c = 10.939(5)$  Å  
 $\alpha = 100.685(7)^\circ$   
 $\beta = 90.650(8)^\circ$   
 $\gamma = 110.307(7)^\circ$  $V = 539.8(4)$  Å<sup>3</sup>  
 $Z = 2$   
 $D_x = 1.459$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
 $\mu = 0.12$  mm<sup>-1</sup>  
 $T = 294(2)$  K  
Block, colourless  
 $0.22 \times 0.20 \times 0.18$  mm



**Figure 1**  
The molecular structure of the title compound, with displacement ellipsoids drawn at the 30% probability level.



**Figure 2**  
A packing diagram of the title compound. Dashed lines indicate C—H...O hydrogen-bond interactions.

#### Data collection

Bruker SMART CCD area-detector diffractometer	2748 measured reflections
$\varphi$ and $\omega$ scans	1897 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1997)	1316 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.974$ , $T_{\max} = 0.979$	$R_{\text{int}} = 0.027$
	$\theta_{\text{max}} = 25.0^\circ$

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.041$   
 $wR(F^2) = 0.126$   
 $S = 1.03$   
 1897 reflections  
 154 parameters  
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0651P)^2 + 0.0629P]$$

where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.15 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.21 \text{ e } \text{\AA}^{-3}$

All H atoms were placed in calculated positions, with C—H = 0.93 and 0.97 Å, and included in the final cycles of refinement using a riding model, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

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#### References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Bruker (1997). SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (1999). SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Dyck, B., Parker, J., Phillips, T., Carter, L., Murphy, B., Summers, R., Hermann, J., Baker, T., Cismowski, M., Saunders, J. & Goodfellow, V. (2003). *Bioorg. Med. Chem. Lett.* **13**, 3793–3796.
- Ehrhardt, H., Mildenerger, H., Sachse, B. & Hartz, P. (1982). Ger. Pat. 3 020 500.
- Konosu, T., Oida, S., Nakamura, Y., Seki, S., Uchida, T., Somada, A., Mori, M., Harada, Y., Kamai, Y., Harasaki, T., Fukuoka, T., Ohya, S., Yasuda, H., Shibayama, T., Inoue, S. I., Nakagawa, A. & Seta, Y. (2001). *Chem. Pharm. Bull.* **49**, 1647–1650.
- Miyauchi, H., Kozuki, K., Tanio, T. & Ohashi, N. (1995). *Bioorg. Med. Chem. Lett.* **5**, 1479–1482.
- Ruel, R., Herpin, T. F., Iben, L., Luo, G., Martel, A., Mason, H., Mattson, G., Poirier, B., Ruediger, E. H., Shi, D., Thibault, C., Yu, G., Zimanyi, I. A., Poindexter, G. S. & Macor, J. E. (2003). *Bioorg. Med. Chem. Lett.* **13**, 4341–4344.
- Sekimata, K., Han, S. Y., Yoneyama, K., Takeuchi, Y., Yoshida, S. & Asami, T. (2002). *J. Agric. Food Chem.* **50**, 3486–3490.
- Sheldrick, G. M. (1997). SADABS (Version 2.0), SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Wan, J., Li, C.-L., Li, X.-M. & Zhang, S.-S. (2005). *Acta Cryst.* **E61**, o307–o308.